Metals and the Fermi Gas

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Electrons in a Metal

Consider electrons in a metal as free non-interacting electrons.

$$\begin{split} \langle n \rangle &= \frac{1}{e^{\beta(\epsilon-\mu)}+1}, \\ \epsilon &= \frac{\hbar^2 k^2}{2m}, \\ g(\epsilon) d\epsilon &= \frac{mV}{\pi^2 \hbar^3} \sqrt{2m\epsilon} d\epsilon. \end{split}$$

Electrons in a Metal

Then, the total number of electrons is

$$N = \int_0^\infty g(\epsilon) \langle n \rangle d\epsilon$$
$$\langle n \rangle = \frac{1}{e^{\beta(\epsilon - \mu)} + 1}.$$

At T = 0 (equivalently $\beta \to \infty$),

- When $\epsilon \leq \mu$, $\langle n \rangle = 1$.
- When $\epsilon > \mu$, $\langle n \rangle = 0$.

Fermi Energy

We define the zero temperature value of chemical potential as Fermi energy ϵ_F . It plays an important role to analyze the (electrical) property of solids. And, we can obtain the total number of particles by using ϵ_F as

$$N = \int_0^\infty g(\epsilon) \langle n \rangle d\epsilon$$

=
$$\int_0^{\epsilon_F} g(\epsilon) d\epsilon$$

=
$$\int_0^{\epsilon_F} \frac{mV}{\pi^2 \hbar^3} \sqrt{2m\epsilon} d\epsilon$$

=
$$\frac{(2\epsilon_F m)^{3/2}}{3\pi^2 \hbar^3} V.$$

This approximation is not accurate in general because we assume independent free particles, but it is surprisingly useful as the simplest example.

Degeneracy Pressure

The Fermi energy is then

$$\epsilon_F = \frac{\hbar^2}{2m} (3\pi^2 N/V)^{2/3} \sim \left(\frac{N}{V}\right)^{2/3}.$$

The energy is

$$E = \int_0^{\epsilon_F} \epsilon g(\epsilon) d\epsilon = \int_0^{\epsilon_F} \epsilon \frac{mV}{\pi^2 \hbar^3} \sqrt{2m\epsilon} d\epsilon$$
$$\sim V \epsilon_F^{5/2} = V \left(\frac{N}{V}\right)^{5/3}.$$

The degeneracy pressure is

$$P \sim \frac{E}{V} = \left(\frac{N}{V}\right)^{5/3} = \rho^{5/3}.$$