

Assignment 4

Byungjoon Min, Statistical Mechanics
(due date: October 15, 2018)

1 Ideal Gases [30 pt]

1.1 Sackur-Tetrode Formula [10 pt]

For the ideal gas, the number of states with two corrected factors $N!$ and h is given by

$$\Omega = V^N \frac{(2\pi mE)^{3N/2}}{\Gamma(3N/2 + 1)N!h^{3N}}. \quad (1)$$

The entropy of the ideal gas is given by $S = k_B \log \Omega$. Combining these two facts, derive the Sackur-Tetrode formula,

$$S(N, V, E) = Nk_B \log \left[\frac{V}{N} \left(\frac{4\pi mE}{3Nh^2} \right)^{3/2} \right] + \frac{5}{2}Nk_B. \quad (2)$$

1.2 Equipartition Theorem [10 pt]

Show that

$$\frac{1}{T} = \left(\frac{\partial S}{\partial E} \right)_{V,N} = \frac{3Nk_B}{2E}. \quad (3)$$

1.3 Equation of state [10 pt]

Show that

$$\frac{P}{T} = \left(\frac{\partial S}{\partial V} \right)_{E,N} = \frac{Nk_B}{V}. \quad (4)$$

2 Harmonic Oscillators [30 pt]

Consider N distinguishable classical harmonic oscillators with mass m and frequency w in one dimension. The Hamiltonian of the oscillators is given by

$$H = \sum_i^N \left[\frac{p_i^2}{2m} + \frac{1}{2}mw^2q_i^2 \right]. \quad (5)$$

2.1 Number of States [10 pt]

The number of states for $H \leq E$ is given by

$$\begin{aligned} \Sigma(E) &= \frac{1}{h^N} \int_{H \leq E} d^N q d^N p \\ &= \frac{1}{h^N} \left(\frac{1}{mw} \right)^N \int_{\sum_i x_i^2 + p_i^2 \leq 2mE} d^N x d^N p \end{aligned}$$

where $x = mwq$. By using the fact that the volume of n -dimensional sphere with radius R is $V_n(R) = R^N \frac{\pi^{n/2}}{(n/2)!}$ Show that

$$\begin{aligned}\Sigma(E) &= \frac{1}{h^N} \left(\frac{1}{mw} \right)^N \frac{\pi^N}{N!} (2mE)^N \\ &= \frac{1}{N!} \left(\frac{E}{\hbar w} \right)^N.\end{aligned}\tag{6}$$

Then, obtain the number of states as

$$\begin{aligned}\Omega(E) &\approx \frac{d\Sigma(E)}{dE} \times E \\ &= \frac{1}{(N-1)!} \left(\frac{E}{\hbar w} \right)^N.\end{aligned}\tag{7}$$

2.2 Entropy [10 pt]

Show that the Entropy of harmonic oscillators

$$\begin{aligned}S &= k_B \log \Omega \\ &= Nk_B \left[1 + \log \left(\frac{E}{N\hbar w} \right) \right].\end{aligned}\tag{8}$$

2.3 Temperature, Pressure, Specific Heat [10 pt]

Show that

$$\begin{aligned}\frac{1}{T} &= \left(\frac{\partial S}{\partial E} \right)_{V,N} = \frac{Nk_B}{E}, \\ \frac{P}{T} &= \left(\frac{\partial S}{\partial V} \right)_{E,N} = 0, \\ C &= \frac{\partial E}{\partial T} = Nk_B.\end{aligned}$$